**BioTecNika Presents** 

## KORGEPT WHEL

**Learn CSIR NET Concepts** 

**Retain Better** 

**Recall Faster** 

Point Wise Notes

Differentiate and Learn



Quick Revision



Memory Retention



Easy Recall

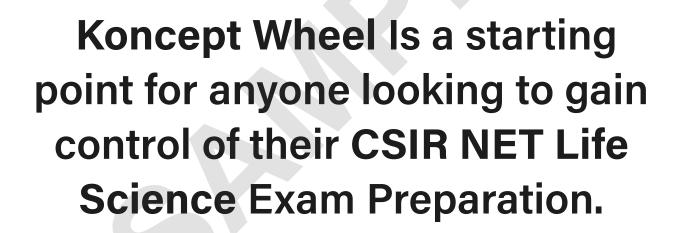
500+ important concepts











- Shekhar Suman CEO MD Biotecnika.



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# KONCEPT WHEEL

**Learn CSIR NET Concepts** 

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**Point Wise Notes** Differentiate and Learn Quick Revision E Easy Recall Memory Retention 500+ important concepts 24/7 chat support





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#### **PREFACE**

Biotecnika, the most trusted study partner for every CSIR NET and GATE aspirant, is committed to making your exam preparation process easy, efficient, and effective. Biotecnika team came up with the unique Konceptika range of products, which precisely aims to do the same. Adding another feather to the cap is Biotecnika's latest study aid-Koncept Wheel.

To excel in CSIR NET Exam, students need to develop a deep understanding of the concepts, prepare brief notes for revisions, and spend an equal amount of time recalling each topic learned. Given the vast nature of the syllabus, aspirants might fall short of time to make detailed notes.

**Koncept Wheel** is a compilation of **500+** important topics from the CSIR **NET** syllabus. These topics are broken down and depicted in the form of a wheel, each spoke of the wheel represents the important definitions, features, examples, etc. These Koncept Wheels will help students understand the concepts easily in a short period of time.

Koncept Wheel is one tool that will change the way you prepare for the exams. With this revolutionary study tool, you can now- Learn quicker, recall faster, and retain longer. By preparing smart with Biotecnika's study tools, stay one step ahead in your exam preparation.

Thank you for choosing Biotecnika's Koncept wheel. We would love to know your feedback, email us at cst@biotecnika.org

sliekhar Suman

CEO & MD - BIOTECNIKA & RASAYANIKA





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Molecules and their Interaction Relevant to Biology

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#### STRUCTURE OF AN ATOM

The number of neutrons defines the isotope of the element.

07

06

The number of protons in the nucleus is the atomic number, defined to which chemical element the atom belongs.

unit of ordinary matter that forms a chemical element.

Atoms are the smallest

01

biotecni (a

O2

Extremely small, typically around 100 picometers across.

The number of electrons is equal to the number of protons.

05

08

Electrons (negative electric charge) present in space around the nucleus.

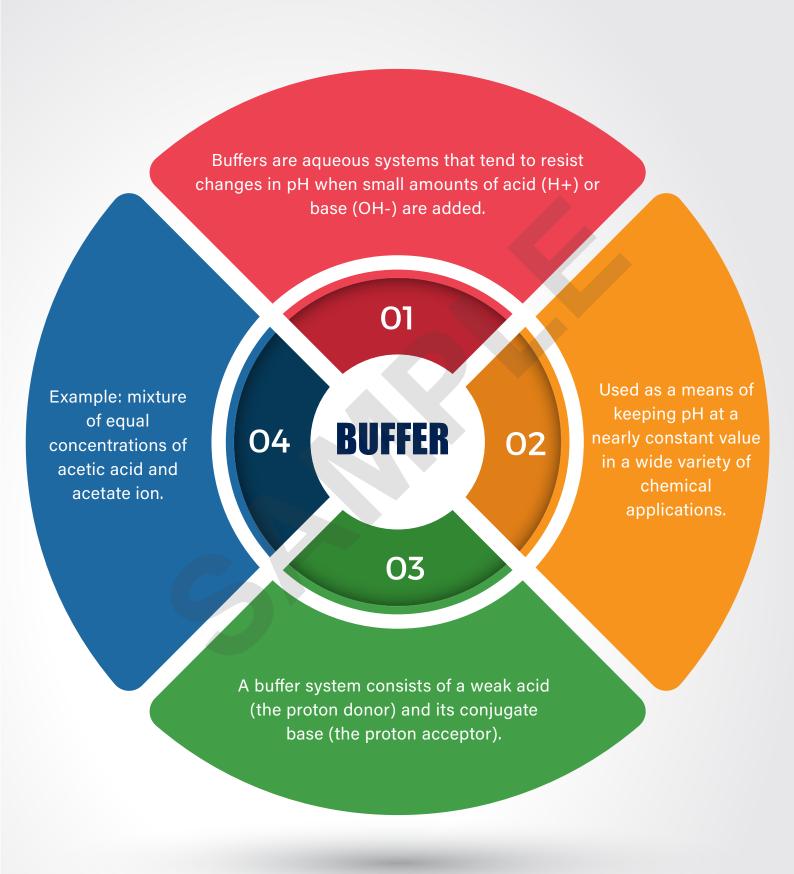
04

03

Composed of a nucleus and one or more electrons bound to the nucleus.

The nucleus is made of a number of protons (positive electric charge) and neutrons (no electric charge).







### **HENDERSON HASSELBALCH EQUATION**





#### **CARBOHYDRATES**

The polysaccharides are sugar polymers The most abundant containing more than 20 biomolecules on Earth. or so monosaccharide units (some have hundreds or thousands of units). It is a biomolecule Oligosaccharides 01 80 consist of short chains of monosaccharide units, or residues. 02 of 2:1 biotecni (a 06 03 Chemical formula Disaccharides, with is  $C_x(H_2O)_{v}$ two monosaccharide 04 05 units. Monosaccharides, or Its three major classes simple sugars, consist of are: monosaccharides, a single polyhydroxy oligosaccharides, and aldehyde or ketone unit. polysaccharides.



 First discovered by Friedrich Miescher. Biopolymers essential Two types of pentose to all known forms of sugar: Ribose in RNA life. and Deoxyribose in acids: Ribonucleic acid DNA, sugar is (RNA) and Deoxyribonunon-planar, non-planarity is called puckering. Pyrimidine (in DNA): Thymine (5-methyl-2,4-dioxypyrimidine) and cytosine (2-oxy-4-aminopyrimidine); (in RNA): Uracil (2,4-dioxypyrimidine) and **NUCLEIC** Cytosine. ACID 08 04 Purines (DNA and RNA): Nucleoside: Sugar and Adenine (6-amino nitrogenous base join to form purine) and Guanine nucleoside (6-oxy-2-aminopurine) The three components Polynucleotides: Polyof Nucleotides: Nitrogemers of nucleotides, nous base (purines and formed by condensation pyrimidines), Five Linkage between two of nucleotides. carbon sugar, ion of nucleotides (5' phosphosphoric acid. phate of one nucleotide known as a phosphodiester bond.

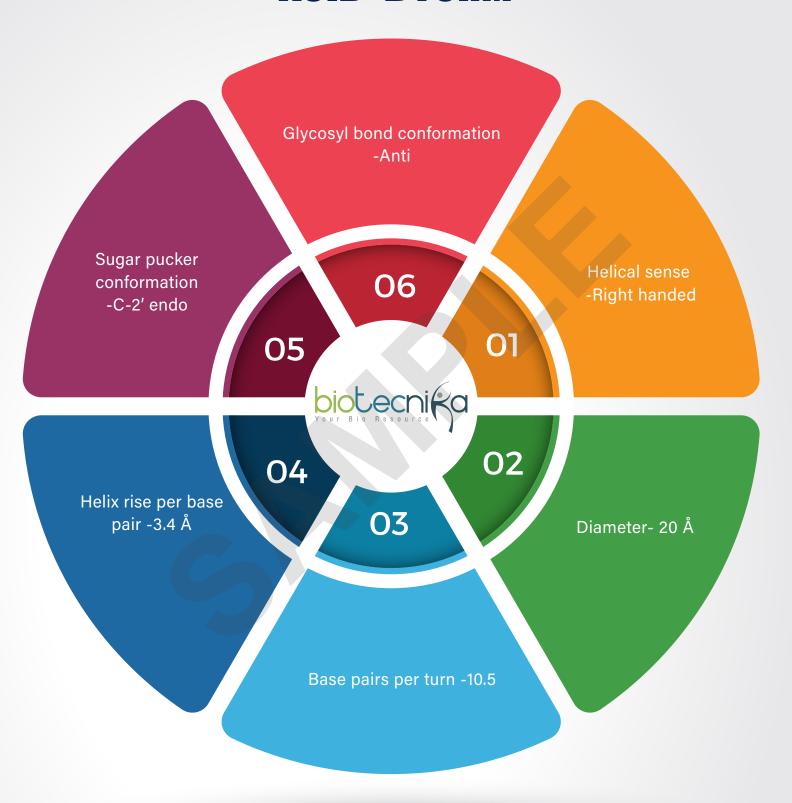


## CONFORMATION OF NUCLEIC ACID-A FORM





## CONFORMATION OF NUCLEIC ACID -B FORM





## CONFORMATION OF NUCLEIC ACID -Z FORM









The most common cross-links are disulfide bonds, formed by the oxidation of a pair of cysteine residues.

07

06

In some proteins, the linear polypeptide chain is cross-linked.

08

Proteins are linear polymers formed by linking the α carboxyl group of one amino acid to the α amino group of another amino acid with a peptide bond (a.k.a amide bond).

01

02

The formation of a peptide bond is accompanied by the loss of a water molecule.

**PROTEINS** 

A polypeptide chain is always written starting with the amino terminal residue.

05

polypeptide
chain has polarity as
its ends are different
(α -amino group at one end
and an α carboxyl group at
the other).

03

04

A series of amino acids joined by peptide bonds form a polypeptide chain, and each amino acid unit in a polypeptide is called a residue.

Most natural polypeptide chains contain between 50 and 2000 amino acid residues.



#### PROTEIN SECONDARY STRUCTURE- α HELIX









# KONCEPT WHEEL

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